[2+3] Cycloadditions of Azomethine Ylides with 1,3-Thiazole-5(4H)thiones

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Dedicated to Professor Max Viscontini on the occasion of his 85th birthday

Thermal reactions of 1,2,3-trisubstituted aziridines 1 with 1,3-thiazole-5(4H)-thiones 6 in toluene yielded, in general, a mixture of two diastereoisomeric spirocyclic [2+3] cycloadducts. The formation of these products can be explained by a stereoselective electrocyclic ring opening of 1 to give an azomethine ylide 2 as the reactive intermediate, which is trapped immediately by 6 via a stereoselective 1,3-dipolar cycloaddition. Only in the case of trans-dimethyl 1-(4-methoxyphenyl)aziridine-2,3-dicarboxylate (trans-1a), four diastereoisomeric cycloadducts were formed (Scheme 4). This result is rationalized by an isomerization of the intermediate azomethine ylide cis-2a to trans-2a.

Introduction. – Azomethine ylides constitute a well-known class of 1,3-dipolar intermediates which were described for the first time by *Huisgen* and co-workers [1][2]. The classical experiments with thermal and photochemical ring opening of the stereoisomeric aziridines *cis*- and *trans*-1a confirm the preservation of orbital symmetry in these processes. With a series of additional experiments, the same authors demonstrated the use of thermally generated azomethine ylides of type 2 (cf. Scheme 2) for the preparation of five-membered heterocycles [3-6]. Simple procedures and the stereoselectivity of aziridine ring opening are attractive features of this concept. An increasing number of similar papers which showed further synthetic application emphasized the potential of this methodology in heterocyclic chemistry [7][8].

The generation of azomethine ylides is, however, not limited to the ring opening of aziridines. Other methods, e.g., desilylation of α -silylonium salts [9], 1,2-prototropic rearrangement of α -amino-acid imines [10], and addition of carbenes or carbenoids onto imines [11] are now frequently used by numerous research groups. In the last few years, further reports have dealt with the use of thermally generated azomethine ylides from corresponding aziridines in syntheses of pyrrolidine and pyrrole derivatives [12]. Especially mentioned should be a recent publication which reports on a new method for the formation of carbapenams and carbapenems by utilizing an azomethine ylide as the reactive intermediate [13]. The same strategy has been used to synthesize penam and penem derivatives [14]. The intermediate azomethine ylide 4 was generated by thermal extrusion of CO_2 from the bicyclic 1,3-oxazolidinone 3 (Scheme 1). Trapping 4 by thiocarbonyl compounds furnished penam derivatives 5 in fair yields.

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Unlike olefinic and acetylenic dipolarophiles, thiocarbonyl compounds have been rarely explored in reactions with azomethine ylides (cf. [7][8][15]), but the results reported by *Gallagher* and co-workers [14] open new horizons in the synthetic use of [2+3] cycloadditions of azomethine ylides with thiocarbonyl compounds.

For many years, our research interest has been focused on different aspects of thiocarbonyl compounds, and their reactivity towards 1,3-dipoles is an essential part of current studies. Thus, reactions with diazo compounds [16], azides [17], thiocarbonyl ylides [18], carbonyl ylides [19], and thiocarbonyl S-imides [20] have been reported. Apart from aliphatic and aromatic thiones, 1,3-thiazole-5(4H)-thiones 6 belong to the most frequently used C=S derivatives in our studies. Very recently, we reported, in preliminary form, on the first example of a 1,3-dipolar cycloaddition of 6 with an azomethine ylide generated thermally from cis-1-methyl-2,3-diphenylaziridine (cis-1b) [21].

In this paper, we present complete results of the thermal reactions of aziridines 1a, 1c, and 1d and 1,3-thiazole-5(4H)-thiones 6a and 6b.

Results and Discussion. – All reactions of aziridines 1 with 1,3-thiazole-5(4*H*)-thiones 6 were carried out in toluene. After complete consumption of 6 (TLC), the solvent was removed and the crude reaction mixture was analyzed by ¹H-NMR spectroscopy. As characteristic signals, located at 6–4 ppm, the absorptions of the H-atoms of the newly formed thiazolidine ring were used to determine the yield and ratio of the isomeric cycloadducts.

As an extension of previous results obtained with cis-1b [21], we used the trans-configured aziridine trans-1c in reactions with 6a and 6b. In both cases, mixtures of two

isomeric products with similar ratio (3:1 and 4:1, resp.) were detected. After chromatographic separation, the crystalline cycloadducts 7 and 8 were obtained (*Scheme 2*). The structures of the pair of diastereoisomers 7a/8a were elucidated by X-ray crystallography (*Fig. 1*). As expected, in both isomers, the Ph groups of the 1,3-thiazolidine ring are *cis*-configured. This result can be rationalized on the basis of a stereoselective ring opening of *trans*-1c yielding the intermediate azomethine ylide 2c. A fast interception by 6a prevents the ylide from isomerization, and, therefore, no 'false isomers' of cycloadducts were formed. The major product 7a is the sterically less hindered one with the Ph groups located opposite to the geminal Me groups.

We propose that the major product of the reaction of *trans*-1c and 6b is again the sterically less hindered 7b. Isomers 7b and 8b show characteristic differences in the chemical shifts $(\Delta \delta)$ of the H-atoms of the 1,3-thiazolidine ring: whereas $\Delta \delta$ in 7b is 0.62 ppm, its value in 8b is 0.79 ppm. Similar differences were observed for 7a $(\Delta \delta = 0.53 \text{ ppm})$ and 8a $(\Delta \delta = 0.90 \text{ ppm})^2$). There is also a remarkable difference in the shifts of the cyclopentane CH₂ groups: the shielding effect of the Ph groups in 8b gives rise to an upfield shift of signals attributed to CH₂ groups of the cyclopentane ring.

Unexpectedly, the reaction of triarylaziridine cis-1d with 6a yielded only a single cycloadduct for which we propose structure 9a (Scheme 3). Reliable criteria for the assignment of the configuration are the chemical shifts of the geminal Me groups of the dihydro-1,3-thiazole ring (Fig. 2): in both isomers with the Ph group at C(9) in a cis-position with respect to the geminal Me groups at C(4), i.e., in 8a (Scheme 2) and 10c [21], the absorptions of $Me_2C(4)$ are shifted towards high field. The chemical shifts observed in 9a correspond very well with those of 9c [21] and exclude the isomeric structure 10a.

In the case of the reaction of *cis*-1d with 6b, the analysis of the crude mixture showed the presence of a predominant product 9b accompanied by traces of a second isomer,

²) In both cases, the isomer of type 8 shows the larger R_f value (TLC, SiO₂).

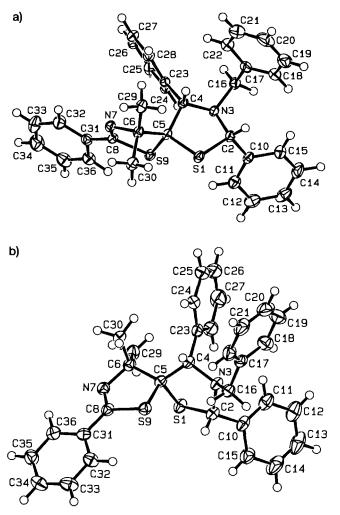


Fig. 1. ORTEP Plots [22] of the molecular structures of a) one of the two independent molecules of 7a and b) 8a (with 50% probability ellipsoids)

most likely 10b. The arguments for the assignment of the structure 9b are analogous to the case of the isomers of type 9 and 10 discussed above.

Finally, we performed reactions of the *cis*- and *trans*-aziridinedicarboxylates, *cis*-1a and *trans*-1a, respectively, with 6a and 6b. Thermal ring opening of *cis*-1a results in the formation of *trans*-2a, as shown by *Huisgen* in his pioneering work [1-3] (*Scheme 4*). Interception with 6a or 6b afforded mixtures of two isomeric cycloadducts in ratios of *ca. 4:1*. In accordance with results presented in *Schemes 2* and 3, we propose the sterically less hindered structures 9d and 9e for the major products and 10d and 10e for the minor ones.

The analogous reaction of **6a** with *trans*-**1a** yielded not only the expected cycloadducts **7d** and **8d**, but also *ca*. 16% of a 3:1 mixture of **9d** and **10d** (*Scheme 4*). This was the only

experiment in which 'false stereoisomers' of 1:1 cycloadducts with 1,3-thiazole-5(4H)-thiones were obtained. The isomers **7d** and **8d** were found in the crude reaction mixture as major products in a ratio of ca. 2:1. By chromatographic workup, the fraction with **7d** and **8d** was separated from the minor components **9d** and **10d**. However, complete

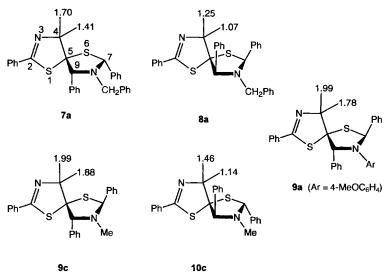


Fig. 2. ¹H-NMR Chemical shifts of 2 Me-C(4) in some spirocyclic 4,5-dihydro-1,3-thiazole derivatives (300 MHz, CDCl₃)

Scheme 4

MeO₂C
$$CO_2Me$$
 Ar
 Ar

separation of **7d** and **8d** could not be achieved using prep. TLC. Therefore, their mixture was used for spectroscopic characterization.

In the case of the reaction of *trans*-1a with 6b, we observed formation of only two cycloadducts, in contrast to the experiment with 6a. The diastereoisomers 7e and 8e were formed in a ratio of *ca.* 3:1. This mixture was separated using prep. TLC, yielding pure products.

In conclusion, the experiments described above show once more that 1,3-thiazole-5(4H)-thiones 6 can be used for efficient trapping of 1,3-dipoles generated as reactive intermediates. Azomethine ylides 2 afforded cycloadducts in good yields, and in almost all cases, mixtures of only two diastereoisomers were formed. Thus, thermal ring opening of aziridines 1 occurs stereoselectively following orbital symmetry rules [23][24]. Azomethine ylides formed in these processes are trapped immediately by 6 without isomerization. The only exception was found in the reaction of trans-1a with 6a. This observation is in accordance with the previously described behavior of trans-1a in reactions with symmetrical C,C-dipolarophiles [3]. Unlike the isomeric cis-1a, trans-1a afforded mix-

tures of 'expected' and 'false' cycloadducts when less reactive dipolarophiles were used 3). The reason for the formation of 'false' cycloadducts 9d and 10d is an isomerization of the azomethine ylide *cis-2a* into *trans-2a* before interception by $6a^4$).

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Experimental Part

General. See [25]. M.p.: in capillary; Büchi-SMP-125 apparatus; uncorrected. IR (KBr, unless otherwise stated): Perkin-Elmer 781. NMR (CDCl₃): Bruker AC 300 (¹H, 300 MHz) and Bruker ARX 300 (¹³C, 75.5 MHz). CI-MS (NH₃): Varian MAT-112S. Elemental analyses were performed by the Mikroanalytisches Laboratorium des Organisch-chemischen Instituts der Universität Zürich.

- 1. Starting Materials. Aziridines 1 were prepared according to known procedures: cis- and trans-dimethyl 1-(4-methoxyphenyl)aziridine-2,3-dicarboxylate (cis- and trans-1a, resp.) [1], trans-1-benzyl-2,3-diphenylaziridine (trans-1c) [6], and cis-1-(4-methoxyphenyl)-2,3-diphenylaziridine (cis-1d) [26]. 1,3-Thiazole-5(4H)-thiones 6a and 6b were prepared according to [27] and [28], resp.
- 2. Reactions of Aziridines 1 with 1,3-Thiazole-5(4H)-thiones 6. 2.1. General Procedure. A stirred soln. of aziridine 1 (1.1 mmol) and 1,3-thiazole-5(4H)-thione 6 (1 mmol) in abs. toluene (1 ml) was heated to 100° (oil bath). After ca. 20 h, the solvent was evaporated, the residue dissolved in CDCl₃, and a weighed amount of 1,1,2,2-tetrachloroethane was added as a standard to establish the yields of the products in the crude mixture by ¹H-NMR spectroscopy (s for 2 H at 5.92 ppm). Then, the soln. was concentrated, and the products were separated by prep. TLC (SiO₂). Anal. pure samples of the products were obtained by recrystallization from alcoholic solvents.
- 2.2. Reaction of trans-1c with 6a. 1 H-NMR: 7a/8a 75:25. (5RS,7RS,9RS)-8-Benzyl-4,4-dimethyl-2,7,9-triphenyl-1,6-dithia-3,8-diazaspiro[4.4]non-2-ene (7a). Isolated as the more polar fraction (hexane/Et₂O 4:1, two-fold development) as a colorless oil which solidified at r.t. (3 d): 360 mg (71%). Recrystallization from MeOH/ CH₂Cl₂ yielded colorless crystals. M.p. 116–118°. IR: 3040m, 2980w, 2860w, 1595s, 1575m (C=N), 1490s, 1450s (br.), 1260s, 1175m, 1075m, 1065m, 1030m, 950s, 910m, 750s, 700s, 690s, 665s. 1 H-NMR: 7.8-6.85 (m, 20 arom. H); 5.03, 4.50 (2s, H-C(7), H-C(9)); 3.66, 3.57 (AB, J = 14.6, CH₂); 1.70, 1.41 (2s, Me₂C). 13 C-NMR: 166.9 (C=N); 140.1, 137.3, 133.9, 133.2 (4s, 4 arom. C); 131.1, 130.7, 130.3, 128.7, 128.4, 128.2, 128.0, 127.9, 127.6, 127.3 (10d, 20 arom. CH); 86.3, 79.5 (2s, C(4), C(5)); 72.8, 66.9 (2d, C(7), C(9)); 51.3 (t, CH₂); 25.8, 22.2 (2q, M₂C). CI-MS: 508 (40), 507 (100, [M + 1] $^+$), 391 (6). Anal. calc. for C₃₂H₃₀N₂S₂ (506.73): C 75.85, H 5.97, N 5.53, S 12.65; found: C 75.96, H 5.94, N 5.46, S 12.37.

Suitable crystals for an X-ray crystal-structure determination were grown from EtOH.

(5RS,7SR,9SR)-8-Benzyl-4,4-dimethyl-2,7,9-triphenyl-1,6-dithia-3,8-diazaspirol 4.4]non-2-ene (8a). Isolated as the less polar fraction (hexane/Et₂O 4:1, twofold development) as a colorless oil which solidified at -20° overnight: 95 mg (19%). Recrystallization from EtOH yielded colorless crystals. M.p. $130-131^\circ$. IR: 3020w, 2800w, 1600m (C=N), 1490m, 1450m, 1260s, 1170m, 1120m (br.), 950s, 775m, 750s, 690vs (br.), 610m. $^1\text{H-NMR}$: 7.8-7.1 (m, 20 arom. H); 5.53, 4.63 (2s, H-C(7), H-C(9)); 3.87, 3.76 (4B, J=13.7, CH_2); 1.25, 1.07 (2s, Me_2C). $^1\text{3}C\text{-NMR}$: 164.0 (s, C=N); 142.0, 137.8, 137.0, 133.6 (4s, 4 arom. C); 131.1, 130.7, 129.3, 128.6, 128.5, 128.4, 128.0, 127.6, 127.2, 127.1 (11d, 20 arom. CH); 92.7, 78.7 (2s, C(4), C(5)); 73.8, 73.0 (2d, C(7), C(9)); 55.4 (t, CH_2); 25.8, 20.3 (2q, Me_2C). CI-MS: 508 (39), 507 (100, $[M+1]^+$). Anal. calc. for $C_{32}H_{30}N_2S_2$ (506.73): C.75.85, H.5.97, N.5.53, S.12.65; found: C.76.11, H.6.32, N.5.42, S.12.40.

Suitable crystals for an X-ray crystal-structure determination were grown from EtOH.

2.3. Reaction of trans-1c with 6b. 1 H-NMR: 7b/8b 80:20. (2RS,4RS,5RS)-3-Benzyl-2,4.12-triphenyl-1,13-dithia-3.11-diazadispiro[4.0.4.3]tridec-11-ene (7b). Isolated as the more polar fraction (hexane/Et₂O 9:1, twofold development) as a viscous, colorless oil: 320 mg (60%). Recrystallization from MeOH yielded colorless crystals. M.p. 170-171°. IR: 2970s, 1595s, 1575s (C=N), 1490s, 1450s (br.), 1310m, 1255s (br.), 1200m, 1120s (br.), 1065s

³) Only ethenetetracarbonitrile and dimethyl acetylenedicarboxylate (dimethyl but-2-ynedioate) exclusively gave 'expected' interception products [3].

⁴⁾ The chromatographically isolated mixture 7d/8d remained unchanged after heating in toluene to 100° for 3 h.

(br.), 1025s, 960vs, 920s, 750vs, 695vs, 660m. $^1\text{H-NMR}$: 7.8-6.85 (m, 20 arom. H); 5.06, 4.44 (2s, H-C(7), H-C(9)); 3.72, 3.58 (AB, J=14.0, $PhCH_2$); 2.55-2.4, 2.25-1.55 (2m, 1:7, 4 CH_2). $^{13}\text{C-NMR}$: 166.7 (s, C=N); 140.6, 136.2, 133.7, 133.3 (4s, 4 arom. C); 131.4, 130.5, 130.3, 128.7, 128.3, 128.1, 127.9, 127.8, 127.7, 127.6, 127.3, 127.1 (12d, 20 arom. CH); 89.8, 84.8 (2s, C(5), C(6)); 74.7, 66.5 (2d, C(2), C(4)); 51.1 (t, $PhCH_2$); 41.2, 31.8, 25.4, 23.9 (4t, 4 CH_2). CI-MS: 534 (40), 533 (100, $<math>[M+1]^+$), 441 (7). Anal. calc. for $C_{34}H_{32}N_2S_2$ (532.77): C 76.65, H 6.06, N 5.26, S 12.04; found: C 76.95, H 6.38, N 5.50, S 12.42.

 $\begin{array}{l} (2\text{RS},4\text{RS},5\text{SR})\text{-}3\text{-}Benzyl\text{-}2,4,12\text{-}triphenyl\text{-}1,13\text{-}dithia\text{-}3,11\text{-}diazadispiro}[4.0.4.3]tridec\text{-}11\text{-}ene & \textbf{(8b)}. \text{ Isolated as the slightly less polar fraction (hexane/Et}_2\text{O 9:1, twofold development) as a viscous, colorless oil: 110 mg (21 %). IR (neat): 2980s, 1590s (C=N), 1480s, 1430s (br.), 1250s (br.), 1205s, 1140s (br.), 1020s (br.), 1000s, 940vs, 760s, 630s. $^{1}\text{H-NMR}$: 7.75-6.9 & (m, 20 arom. H); 5.38, 4.59 & (2s, H-C(7), H-C(9)); 3.93, 3.78 & (AB, J=14.9, PhCH}_2); 2.65-2.5, 2.1-1.0 & (2m, 1:7, 4 CH}_2). $^{13}\text{C-NMR}$: 163.6 & (s, C=N); 139.0, 138.1, 135.2, 133.9 & (4s, 4 arom. C); 130.8, 130.5, 129.8, 128.7, 128.5, 128.2, 127.9, 127.8, 127.7, 127.4, 127.2 & (11d, 20 arom. CH); 90.2, 89.4 & (2s, C(5), C(6)); 77.0, 70.9 & (2d, C(2), C(4)); 53.7 & (t, PhCH}_2); 38.9, 33.4, 25.3, 25.0 & (4t, 4 CH}_2). CI-MS: 534 & (35), 533 & (100, [M+1]^+). Anal. calc. for $C_{34}\text{H}_{32}\text{N}_2\text{S}_2$ & (532.77): C 76.65, H 6.06, N 5.26, S 12.04; found: C 76.93, H 5.78, N 5.45, S 11.79. \\ \end{array}$

- 2.4. Reaction of cis-1d with 6a. 1 H-NMR: Only 9a was detected. Chromatography (hexane/Et₂O 4:1) yielded 410 mg (79%) of (5RS,7SR,9RS)-8-(4-methoxyphenyl)-4.4-dimethyl-2.7,9-triphenyl-1.6-dithia-3,8-diazaspiro-[4.4]non-2-ene (9a) as a viscous, colorless oil. Crystallization from EtOH gave colorless crystals. M.p. 84-86°. IR: 2980w, 2930w, 1595m (C=N), 1510vs, 1450m, 1245vs, 1170m, 1035m, 955s, 825m, 765m, 695s. 1 H-NMR: 7.6-7.55 (m, 5 arom. H); 7.4-7.15 (m, 10 arom. H); 6.57, 6.49 (AA'BB', J = 8.9, 4 arom. H); 6.14, 5.66 (2s, H-C(7), H-C(9)); 3.57 (s, MeO); 1.99, 1.78 (2s, Me₂C). 13 C-NMR: 164.7 (s, C=N); 154.4, 138.7, 138.3, 137.9, 133.3 (5s, 5 arom. C); 131.0, 128.9, 128.4, 128.2, 127.9, 127.8, 127.5, 123.2, 113.7 (9d, 19 arom. CH); 85.2, 82.0 (2s, C(5), C(6)); 75.9, 68.0 (2d, C(7), C(9)); 55.1 (q, MeO); 26.4, 23.1 (2q, Me₂C). CI-MS: 524 (36), 523 (100, [M+1]⁺), 280 (16), 244 (12). Anal. calc. for C₃₂H₃₀N₂OS₂ (522.73): C 73.53, H 5.78, N 5.36, S 12.27; found: C 73.18, H 5.61, N 5.28, S 11.95.
- 2.5. Reaction of cis-1d with 6b. 1 H-NMR: 9b/10b 97:3. After chromatographic workup (hexane/Et $_2$ O 17:3), only the major product could be isolated as a viscous, colorless oil: 370 mg (68%) of (2RS,4SR,5SR)-3-(4-methoxyphenyl)-2,4,12-triphenyl-1,13-dithia-3,11-diazadispiro[4.0.4.3]tridec-11-ene (9b). Crystallization from MeOH at -20° yielded colorless crystals. M.p. 136–138°. IR: 2950m, 1590m, 1570m (C=N), 1510m, 1455s, 1240vs, 1180m, 1040s, 945m, 920m, 825m, 760s, 695vs. 1 H-NMR: 7.6–7.15 (m, 15 arom. H); 6.59, 6.51 (AA'BB', J = 9.0, 4 arom. H); 6.24, 5.70 (2s, H-C(7), H-C(9)); 3.60 (s, MeO); 3.05–2.9, 2.7–2.5, 2.3–1.8 (3m, 1:2:5, 4 CH $_2$). 1 3C-NMR: 165.1 (C=N); 154.1, 139.1, 138.1, 138.0, 133.4 (5s, 5 arom. C); 130.9, 129.3, 128.4, 128.2, 127.9, 127.8, 127.7, 127.6, 123.1, 113.5 (10d, 19 arom. CH); 92.9, 83.5 (2s, C(5), C(6)); 76.6, 68.2 (2d, C(2), C(4)); 55.0 (q, MeO); 39.0, 32.5, 25.5, 24.3 (4t, 4 CH $_2$). CI-MS: 550 (35), 549 (100, [M + 1] $^+$), 302 (11), 248 (11), 244 (17), 212 (16). Anal. calc. for $C_{34}H_{32}N_2OS_2$ (548.77): C 74.41, H 5.88, N 5.11, S 11.68; found: C 74.69, H 5.95, N 5.03, S 11.47.
- 2.6. Reaction of cis-1a with 6a. 1 H-NMR: 9d/10d 80:20. Dimethyl (5RS, 7SR, 9RS)-8-(4-Methoxyphenyl)-4,4-dimethyl-2-phenyl-1,6-dithia-3,8-diazaspirof 4.4]non-2-ene-7,9-dicarboxylate (9d). Isolated as the less polar fraction (hexane/Et₂O 4:1, twofold development) as a viscous, colorless oil: 230 mg (47%). Crystallization from EtOH gave colorless crystals. M.p. $161-162^\circ$. IR: 2950w, 1760m, 1735vs (br., C=O), 1590w, 1575w (C=N), 1515vs (br.), 1445m, 1435m, 1250vs (br.), 1200s, 1170vs, 1040s, 1010s, 955s, 820s, 770s. 1 H-NMR: 7.75-7.7 (m, 2 arom. H); 7.5-7.35 (m, 3 arom. H); 6.8-6.7 (m, 4 arom. H); 5.63, 5.34 (2s, H-C(7), H-C(9)); 3.75, 3.72, 3.61 (3s, 3 MeO); 1.88, 1.54 (2s, Me₂C). 13 C-NMR: 170.2, 169.9 (2s, 2 C=O); 164.0 (s, C=N); 154.6, 137.5, 132.9 (3s, 3 arom. C); 131.5, 128.5, 127.9, 118.5, 114.8 (5d, 9 arom. CH); 82.0, 79.9 (2s, C(4), C(5)); 72.6, 63.3 (2d, C(7), C(9)); 55.4, 52.7, 51.7 (3q, 3 MeO); 24.7, 21.7 (2q, Me₂C). CI-MS: 488 (31), 487 (100, [M+1]+). Anal. calc. for $C_{24}H_{26}N_2O_5S_2$ (486.61): C 59.24, H 5.39, N 5.76, S 13.18; found: C 58.91, H 5.50, N 5.45, S 13.07.

Dimethyl (5RS,7RS,9SR)-8-(4-Methoxyphenyl)-4,4-dimethyl-2-phenyl-1,6-dithia-3,8-diazaspiro[4,4]non-2-ene-7,9-dicarboxylate (10d). Isolated as the slightly more polar fraction (hexane/Et₂O 4:1, twofold development) as a viscous, colorless oil: 63 mg (13%). Crystallization from EtOH gave colorless crystals. M.p. 157-159°. IR: 2950m, 1770vs (C=O), 1735vs (C=O), 1595m, 1575m (C=N), 1515vs (br.), 1450m, 1435m, 1330m, 1260vs (br.), 1160vs (br.), 1040s, 955s, 815s, 765s, 680s. 1 H-NMR: 7.8-7.75 (m, 2 arom. H); 7.45-7.35 (m, 3 arom. H); 6.83, 6.69 (AA'BB', J = 9.0, 4 arom. H); 5.53, 5.21 (2s, H-C(7), H-C(9)); 3.80, 3.75, 3.65 (3s, 3 MeO); 1.61, 1.56 (2s, Me₂C). 13 C-NMR: 170.4, 169.4 (2s, 2 C=O); 164.1 (s, C=N); 154.2, 137.7, 133.2 (3s, 3 arom. C); 131.3, 128.4, 128.0, 117.3, 114.8 (5d, 9 arom. CH); 87.6, 77.2 (2s, C(4), C(5)); 72.7, 63.0 (2d, C(2), C(9)); 55.4, 52.8, 52.2 (3q, 3 MeO); 25.2, 21.2 (2q, Me₂C). CI-MS: 488 (27), 487 (100, [M+1]+), 262 (11), 226 (8). Anal. calc. for $C_{24}H_{26}N_{2}O_{3}S_{2}$ (486.61): C 59.24, H 5.39, N 5.76, S 13.18; found: C 59.14, H 5.43, N 5.72, S 13.23.

2.7. Reaction of cis-1a with 6b. 1 H-NMR: 9e/10e 85:15. Dimethyl (2RS,4SR,5SR)-3-(4-Methoxyphenyl)-12-phenyl-1,13-dithia-3,11-diazadispiro[4.0.4.3]tridec-11-ene-2,4-dicarboxylate (9e). Isolated as the less polar fraction (hexane/Et₂O 17:3) as a viscous, colorless oil which solidified at r.t.: 380 mg (74%). Crystallization from EtOH with small amounts of CH₂Cl₂ gave colorless crystals. M.p. 118–120°. IR: 2950m, 1755vs (br., C=O), 1595vs, 1575m (C=N), 1510vs, 1450s, 1435s, 1335s, 1270vs, 1250vs, 1195vs, 1170vs, 1035m, 985m, 950m, 900m, 815s, 770s, 690m. 1 H-NMR: 7.65–7.6 (m, 2 arom. H); 7.35–7.25 (m, 3 arom. H); 6.66, 6.57 (AA'BB', J = 9.4, 4 arom. H); 5.67, 5.29 (2s, H-C(2), H-C(4)); 3.74, 3.71, 3.57 (3s, 3 MeO); 2.8–1.7 (m, 4 CH₂). 13 C-NMR: 170.1, 169.8 (2s, 2 C=O); 165.0 (s, C=N); 154.6, 137.5, 133.2 (3s, 3 arom. C); 131.4, 128.5, 128.0, 118.6, 114.7 (5d, 9 arom. CH); 93.0, 78.9 (2s, C(5), C(6)); 74.0, 63.4 (2d, C(2), C(4)); 55.4, 52.8, 51.6 (3q, 3 MeO); 38.1, 30.9, 25.2, 24.4 (4t, 4 CH₂). CI-MS: 514 (2s). 513 (100, [M + 1] $^+$), 288 (7), 248 (9). Anal. calc. for C₂ $_6$ H₂₈N₂O₅S₂ (512.65): C 60.91, H 5.51, N 5.47, S 12.51; found: C 60.59, H 5.87, N 5.28, S 12.33.

Dimethyl (2RS,4SR,5RS)-3-(4-Methoxyphenyl)-12-phenyl-1,13-dithia-3,11-diazadispiro[4.0.4.3] tridec-11-ene-2,4-dicarboxylate (10e). Isolated as the slightly less polar fraction (hexane/Et₂O 17:3) as a viscous, colorless oil, contaminated with ca. 20% of $9e^{-5}$). 1 H-NMR: 7.85-7.8 (m, 2 arom. H); 7.45-7.35 (m, 3 arom. H); 6.81, 6.68 (AA'BB', J=9.1, 4 arom. H); 5.52, 5.22 (2s, H-C(2), H-C(4)); 3.80, 3.73, 3.61 (3s, 3 MeO); 2.3-1.7 (m, 4 CH₂). 13 C-NMR: 170.7, 169.5 (2s, 2 C=O); 165.0 (s, C=N); 154.2, 137.7, 133.3 (3s, 3 arom. C); 131.4, 128.4, 128.0, 117.3, 114.9 (5d, 9 arom. CH); 87.7, 85.3 (2s, C(5), C(6)); 73.8, 63.1 (2d, C(2), C(4)); 55.4, 52.9, 52.2 (3q, 3 MeO); 37.6, 34.0, 25.0, 24.5 (4t, 4 CH₂).

2.8. Reaction of trans-1a with 6b. 1 H-NMR: 7e/8e 75:25. Dimethyl (2RS,4RS,5RS)-3-(4-Methoxyphenyl)-12-phenyl-1.13-dithia-3.11-diazadispiro[4.0.4.3]tridec-11-ene-2.4-dicarboxylate (7e). Isolated as the more polar fraction (hexane/Et₂O 4:1) as a viscous, colorless oil: 280 mg (55%). IR (neat): 2950m, 1740vs (br., C=O), 1595vc, 1575vc (C=N), 1510vs, 1445m, 1430m, 1250vs (br.), 1195vs, 1165vs, 1035m, 810m, 765m 1 H-NMR: 7.8-7.7 (m, 2 arom. H); 7.45-7.35 (m, 3 arom. H); 6.85, 6.67 (AA'BB', J = 9.1, 4 arom. H); 5.44, 5.04 (2v, H-C(2), H-C(4)); 3.79, 3.73, 3.60 (3v, 3 MeO); 2.35-1.65 (v, 4 CH₂). 13 C-NMR: 169.7, 169.5 (2v, 2 C=O); 164.9 (v, C=N); 153.8, 138.4, 133.2 (3v, 3 arom. C); 131.5, 128.5, 128.1, 115.6, 114.9 (5v, 9 arom. CH); 92.7, 80.3 (2v, C(5), C(6)); 72.6, 64.7 (2v, C(2), C(4)); 55.5, 52.9, 52.1 (3v, 3 MeO); 38.4, 32.7, 25.0, 24.1 (v, 4 CH₂). CI-MS: 514 (31), 513 (100, [v + 1]v), 391 (7), 288 (19), 226 (10). Anal. calc. for C₂₆H₂₈N₂O₅S₂ (512.65); C 60.91, H 5.51, N 5.47, S 12.51; found: C 60.94, H 5.65, N 5.21, S 12.43.

Dimethyl (2RS,4RS,5SR)-3-(4-Methoxyphenyl)-12-phenyl-1,13-dithia-3,11-diazadispiro[4.0.4.3] tridec-11-ene-2,4-dicarboxylate (8e). Isolated as the slightly less polar fraction (hexane/Et₂O 4:1) as a viscous, colorless oil: 85 mg (17%). IR (CHCl₃): 2960m, 1750vs (C=O), 1730vs (C=O), 1595w, 1575w (C=N), 1510vs, 1435m, 1250vs, 1170s, 1040m, 975m. H-NMR: 7.75-7.7 (m, 2 arom. H); 7.45-7.35 (m, 3 arom. H); 6.77, 6.75 (AA'BB', J = 9.3, 4 arom. H); 5.67, 5.11 (2s, H-C(2), H-C(4)); 3.76, 3.75 (2s, 1:2, 3 MeO); 2.35-1.75 (m, 4 CH₂). ¹³C-NMR: 169.2, 169.0 (2s, 2 C=O); 162.7 (s, C=N); 153.7, 138.5, 133.3 (3s, 3 arom. C); 131.4, 128.1, 126.1, 115.4, 115.0 (5d, 9 arom. CH); 88.4, 87.9 (2s, C(5), C(6)); 71.5, 64.1 (2d, C(2), C(4)); 55.1, 52.9, 52.5 (3s, 3 MeO); 36.7, 35.4, 25.1, 25.0 (4t, 4 CH₂). CI-MS: 514 (31), 513 (100, [M + 1] $^+$), 391 (9), 376 (9), 288 (21), 248 (7), 226 (12), 138 (6). Anal. calc. for C₂6H₂8N₂O₅S₂ (512.65): C 60.91, H 5.51, N 5.47, S 12.51; found: C 60.53, H 5.75, N 5.00, S 12.48.

2.9. Reaction of trans-1a with 6a. A stirred soln. of trans-1a (291 mg, 1.1 mmol) and 6a (221 mg, 1 mmol) in toluene (1 ml) was heated to 80° for 11 h. After that time, no trans-1a was present in the mixture (TLC). After evaporation, the ¹H-NMR revealed the presence of 7a/8d/9d/10d 58:26:12:4 in a total yield of 67% (1,1,2,2-te-trachloroethane as standard). Prep. TLC (SiO₂, hexane/Et₂O 3:2) afforded 260 mg (54%) of 7d/8d 2:1 which could not be separated ⁶). The minor components 9d and 10d were isolated as pure compounds (34 mg (7%) and 10 mg (2%), resp.).

7d: ¹H-NMR: 7.80 – 7.77 (m, 2 arom. H); 7.47 – 7.37 (m, 3 arom. H); 6.84, 6.80 (AA'BB', J = 9.1, 4 arom. H); 5.41, 5.14 (2s, H–C(2), H–C(4)); 3.78, 3.75, 3.67 (3s, 3 MeO); 1.56, 1.49 (2s, Me₂C). ¹³C-NMR: 169.6, 169.5 (2s, 2 C=O); 165.6 (s, C=N); 153.7, 138.2, 132.8 (3s, 3 arom. C); 131.8, 128.6, 128.3, 115.3, 115.0 (5d, 9 arom. CH); 81.8, 81.1 (2s, C(5), C(6)); 70.4, 64.0 (2d, C(2), C(4)); 55.6, 52.9, 52.3 (3q, 3 MeO); 24.8, 22.7 (2q, Me_2 C).

8d: ¹H-NMR: 7.74–7.71 (m, 2 arom. H); 7.47–7.37 (m, 3 arom. H); 6.88, 6.68 (AA'BB', J = 9.1, 4 arom. H); 5.67, 5.08 (2s, H–C(2), H–C(4)); 3.80, 3.77, 3.76 (3s, 3 MeO); 1.51, 1.75 (2s, Me₂C). ¹³C-NMR: 169.2, 168.8 (2s, 2 C=O); 163.0 (s, C=N); 153.5, 138.2, 132.7 (3s, 3 arom. C); 131.6, 128.6, 128.3, 115.0, 114.9

⁵⁾ Even after repeated chromatography (prep. TLC), 10e could not be obtained in pure form. Therefore, the NMR data stem from a sample of 10e containing ca. 20% of 9e.

⁶⁾ The NMR data of 7d and 8d are taken from a spectrum of the 2:1 mixture.

(5d, 9 arom. CH); 104.1, 90.6 (2s, C(5), C(6)); 70.7, 63.8 (2d, C(2), C(4)); 55.6, 52.9, 52.4 (3q, 3 MeO); 26.0, 21.7 (2q, Me₂C).

Data of 7d/8d 2:1: IR (neat): 3000m, 2950m, 1750vs (br.), 1515vs, 1445s, 1260vs, 1175vs, 1040s, 955s, 810m, 690s. EI-MS: 486 (22, M^+), 427 (47, $[M-CO_2Me]^+$), 324 (11), 265 (12), 206 (26), 145 (100, $[Ph-C\equiv N-CMe_2]^+$), 134 (23), 104 (29). Anal. calc. for $C_{24}H_{26}N_2O_5S_2$ (486.61): C 59.24, H 5.39, N 5.76, S 13.18; found: C 58.58, H 5.21, N 5.65, S 13.10.

3. Crystal-Structure Determination of Compounds 7a and 8a (see Table and Fig. 1)⁷). All measurements were made on a Rigaku-AFC5R diffractometer in the $\omega/2\theta$ -scan mode using graphite-monochromated MoK, radiation ($\dot{\lambda}=0.71069$ Å) and a 12-kW rotating anode generator. The intensities were corrected for Lorentz and polarization effects, but not for absorption. Data collection and refinement parameters are listed in the Table, views of the molecules are shown in Fig. 1. The structures were solved by direct methods using SHELXS86 [29], which revealed the positions of all non-H-atoms. In the case of 7a, there are two independent molecules in the asymmetric

Table. Crystallographic Data of 7a and 8a

	7 a	8a
Crystallized from	EtOH	EtOH
Empirical formula	$C_{32}H_{30}N_2S_2$	$C_{32}H_{30}N_2S_2$
M_r	506.72	506.72
Crystal color, habit	colorless, prism	colorless, prism
Crystal dimensions [mm]	$0.18 \times 0.35 \times 0.44$	$0.23 \times 0.28 \times 0.47$
Temperature [K]	173(1)	173(1)
Crystal system	triclinic	monoclinic
Space group	$P\overline{1}$	$P2_1/c$
Z	4	4
Reflections for cell determination	25	25
2θ Range for cell determination [*]	37-40	24-26
Unit cell parameters a[Å]	14.124(2)	14.073(3)
b [Å]	16.363(2)	9.378(3)
$c\left[\mathbf{\mathring{A}}\right]$	11.676(2)	21.664(3)
α [°]	90.74(1)	90
β [°]	103.29(1)	105.78(1)
γ["]	96.12(1)	90
$V[Å^3]$	2609.4(6)	2751(1)
$D_{\rm x}$ [g cm ⁻³]	1.290	1.220
$\mu(MoK_2)$ [mm ⁻¹]	0.228	0.216
$2\theta_{(\max)}$ [°]	55	55
Total reflections measured	12452	6989
Symmetry independent reflections	11968	6326
Reflections used $[I > 2\sigma(I)]$	8457	4471
Parameters refined	889	445
Final R	0.0418	0.0422
$wR (w = [\sigma^2(F_0) + (0.005F_0)^2]^{-1})$	0.0385	0.0384
Goodness of fit	1.484	1.563
Final $\Delta_{\text{max}}/\sigma$	0.0006	0.0005
$\Delta \rho$ (max; min) [e Å ⁻³]	0.31; -0.27	0.27; -0.22
Range of $\sigma(d(C-C))$ [Å]	0.003 - 0.004	0.003 - 0.005

⁷⁾ Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as deposition No. CCDC-100927. Copies of the data can be obtained, free of charge, on application to the CCD, 12 Union Road, Cambridge CB2 1EZ, U.K. (fax: + 44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.uk).

unit. The atomic coordinates of the two molecules were tested carefully for a relationship from a higher symmetry space group using the MISSYM routine [30] of the program PLATON [31], but none could be found. The non-H-atoms were refined anisotropically. All of the H-atoms of 7a and 8a were located in difference electron density maps, and their positions were allowed to refine together with individual isotropic displacement parameters. All refinements were carried out on F using full-matrix least-squares procedures. Corrections for secondary extinction were not applied. Neutral-atom scattering factors for non-H-atoms were taken from [32a] and the scattering factors for H-atoms from [33]. Anomalous dispersion effects were included in F_{calc} [34]; the values for f and f were those of [32b]. All calculations were performed using the TEXSAN crystallographic software package [35].

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